Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Kandasamy Chinnakali,^a Thangavel Ravishankar,^b Kamaraj Sriraghavan,^c Ibrahim Abdul Razak,^d Hoong-Kun Fun,^d* Suchada Chantrapromma^dt and Vayalakkavoor T. Ramakrishnan^c

^aDepartment of Physics, Anna University, Chennai 600025, India, ^bDepartment of Physics, Deen Dayal Engineering College, Kunnavalam 600210, Tamil Nadu, India, ^cDepartment of Organic Chemistry, University of Madras, Guindy Campus, Chennai 600025, India, and ^dX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

+ Permanent address: Department of Chemistry, Faculty of Science, Prince of Songkhla University, Hat-Yai, Songkhla 90112, Thailand.

Correspondence e-mail: hkfun@usm.my

Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(C-C) = 0.005 \text{ Å}$ R factor = 0.078 wR factor = 0.223 Data-to-parameter ratio = 13.6

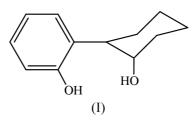
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The crystal structure of the title compound, $C_{12}H_{16}O_2$, contains two crystallographically independent molecules in the asymmetric unit. The cyclohexane rings of these two molecules adopt chair conformations with the hydroxyl groups and benzene rings equatorially attached. The crystal structure is stabilized by $O-H\cdots O$ hydrogen bonds and weak $C-H\cdots \pi$ interactions involving the phenyl rings.

Comment

Cyclohexanol and its derivatives have proven to be important tools in both biochemical and physiological studies of the cholimergin nerve terminal (Rogers *et al.*, 1989). Also, many of the cyclohexanol derivatives exhibit good receptor properties against the inhibitor of acetylcholine storage by nerve terminal synaptic vesicles (Marshall & Parsons, 1987). The crystal structure determination of the title compound, (I), one of the above derivatives, was performed in order to elucidate its molecular conformation.



The asymmetric unit of (I) contains two crystallographically independent molecules linked by an $O1B-H1C\cdots O2A$ hydrogen bond, with their centroid at (0.252, 0.250, 0.497). The corresponding bond lengths and angles of these two molecules agree with each other and show normal values. The cyclohexane ring in both molecules adopts the chair conformation and the hydroxyl and phenyl groups are equatorially attached. An O-H···O intramolecular hydrogen bond is observed in each of the two molecules in the asymmetric unit (Table 1). In the solid state, the two independent molecules are alternately linked by $O-H \cdots O$ hydrogen bonds to form an infinite onedimensional chain along the *a* direction. The crystal structure is further stabilized by a number of weak $C-H \cdots \pi$ interactions involving the phenyl rings of molecule A (πPA = centroid of C1A–C6A) and molecule B (πPB = centroid of C1B-C6B).

Experimental

To a stirred suspension of magnesium turnings (0.75 g-atom) in dry THF (25 ml) under a nitrogen atmosphere was added dropwise a

Received 30 April 2001 Accepted 25 June 2001 Online 29 June 2001

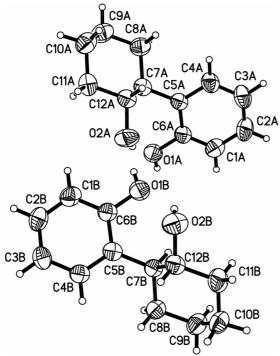


Figure 1

The structure of (I) showing 50% probability displacement ellipsoids and the atom-numbering scheme.

solution of 2-bromoanisole (3.8 ml, 0.03 M) in dry THF (15 ml). After the addition was complete, the solution was cooled to 263 K (using an ice-salt mixture), and cuprous iodide (0.29 g, 1.5 mmol) was added. The resulting mixture was then stirred for a further 15 min, after which time cyclohexane oxide (2 ml, 0.02 M) in dry THF (10 ml) was added dropwise. After completion of the reaction, it was quenched with a saturated solution of ammonium sulfate (50 ml). The reaction mixture was then extracted with ethyl acetate and the organic layer was washed with water, then dried over anhydrous MgSO₄. After removal of the solvent, the residue was chromatographed over silica gel to afford trans-2-(2-methoxyphenyl)cyclohexanol (4.1 g, 98%) as a viscous liquid in 98% yield. It was then further demethylated with chlorotrimethylsilane (2.17 g, 0.02 M) and sodium iodide (3 g, 0.02 M)0.02 M) using dry acetonitrile as solvent, affording the title compound as a crystalline solid (m.p. 367-369 K).

Crystal data

$C_{12}H_{16}O_2$	Z = 4
$M_r = 192.25$	$D_x = 1.224 \text{ Mg m}^{-3}$
Triclinic, P1	Mo $K\alpha$ radiation
a = 9.0290 (7) Å	Cell parameters from 2614
b = 10.0804 (8) Å	reflections
c = 12.1751 (10) Å	$ heta=1.8 extrm{-}28.4^\circ$
$\alpha = 72.031 \ (2)^{\circ}$	$\mu = 0.08 \text{ mm}^{-1}$
$\beta = 81.913 \ (2)^{\circ}$	T = 293 (2) K
$\gamma = 89.545 \ (2)^{\circ}$	Plate, colourless
$V = 1042.9 (1) \text{ Å}^3$	$0.48 \times 0.40 \times 0.14 \text{ mm}$

Data collection

Siemens SMART CCD area- detector diffractometer	$\begin{aligned} R_{\rm int} &= 0.046\\ \theta_{\rm max} &= 25.0^\circ \end{aligned}$
ω scans	$h = -10 \rightarrow 10$
5444 measured reflections	$k = -11 \rightarrow 11$
3503 independent reflections	$l = 0 \rightarrow 14$
1965 reflections with $I > 2\sigma(I)$	

Refinement

Refinement on F^2 H-atom parameters constrained $w = 1/[\sigma^{\frac{1}{2}}(F_o^2) + (0.1210P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $R[F^2 > 2\sigma(F^2)] = 0.078$ $wR(F^2) = 0.223$ S = 0.96 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$ 3503 reflections $\Delta \rho_{\rm min} = -0.32 \text{ e } \text{\AA}^{-3}$ 257 parameters

Table 1

Hydrogen-bonding geometry (Å, °).

 πPA is the centroid of ring C1A-C6A and πPB is the centroid of ring C1B-C6B.

$D-\mathrm{H}\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O1A - H1A \cdots O2B^{i}$	0.82	1.87	2.655 (4)	159
$O1B-H1C\cdots O2A$	0.82	1.86	2.661 (4)	167
$O2A - H2A \cdots O1A$	0.82	2.01	2.803 (4)	163
$O2B - H2C \cdot \cdot \cdot O1B$	0.82	2.11	2.817 (4)	144
$C9A - H9B \cdots \pi PB^{ii}$	0.97	3.11	3.965 (5)	148
$C9B-H9D\cdots\pi PA^{iii}$	0.97	3.06	3.924 (4)	149
$C11A - H11B \cdots \pi PB^{iv}$	0.97	3.09	3.931 (4)	146
$C11B - H11C \cdots \pi PA^{v}$	0.97	3.11	3.930 (4)	144

Symmetry codes: (i) 1 + x, y, z; (ii) 1 - x, -y, 1 - z; (iii) 1 - x, 1 - y, 1 - z; (iv) -x, -y, 1-z; (v) -x, 1-y, 1-z.

After checking their presence in the difference map, all the H atoms were placed in geometrically calculated positions and a riding model was used for their refinement. A rotating group refinement was used for the hydroxyl groups.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 1990).

The authors would like to thank the Malaysian Government and Universiti Sains Malaysia for research grant R&D No. 305/PFIZIK/610942.

References

Bruker (1998). SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.

Marshall, I. G. & Parsons, S. M. (1987). Trends Neurosci. 10, 174.

Rogers, G. A., Parson, S. M., Anderson, D. C., Nilsson, L. M., Batir, B. A., Kornreich, W. D., Kaufman, R., Jacobs, R. S. & Kirtman, B. (1989). J. Med. Chem. 32, 1217-1230.

Sheldrick, G. M. (1997). SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.

Spek, A. L. (1990). Acta Cryst. A46, C-34.